

How diffusion is
affected by
different shaped
containers.

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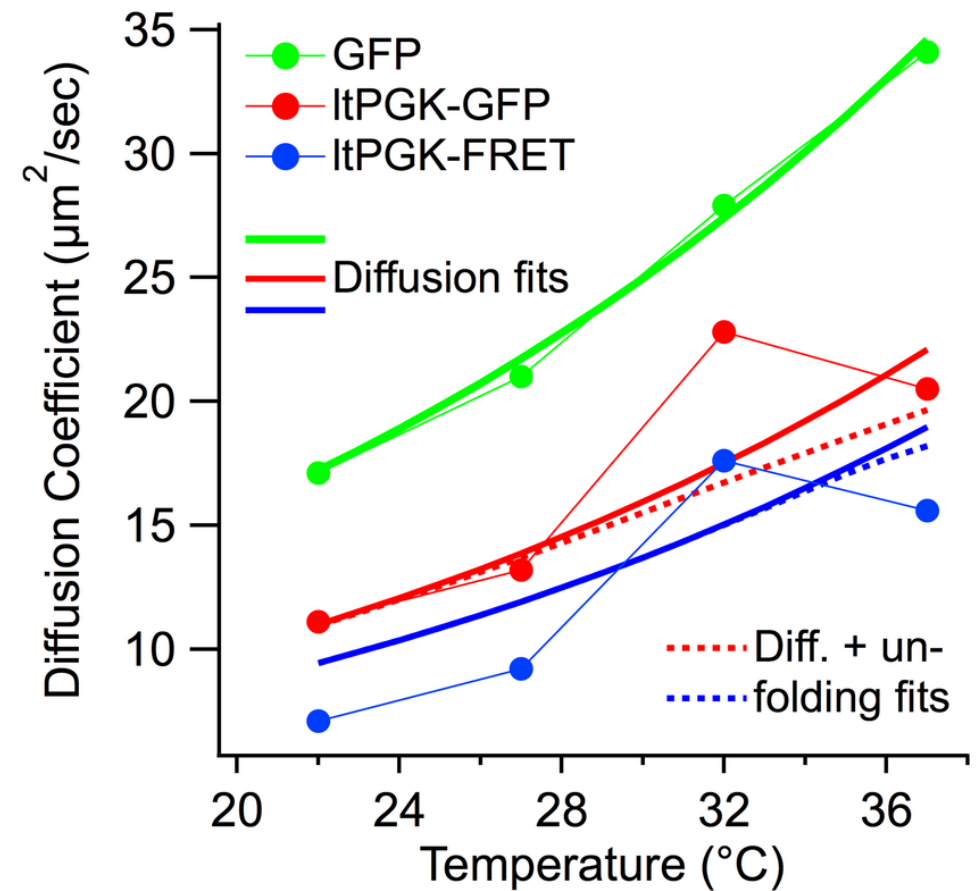


Introduction - Background

The rate of diffusion depends on:

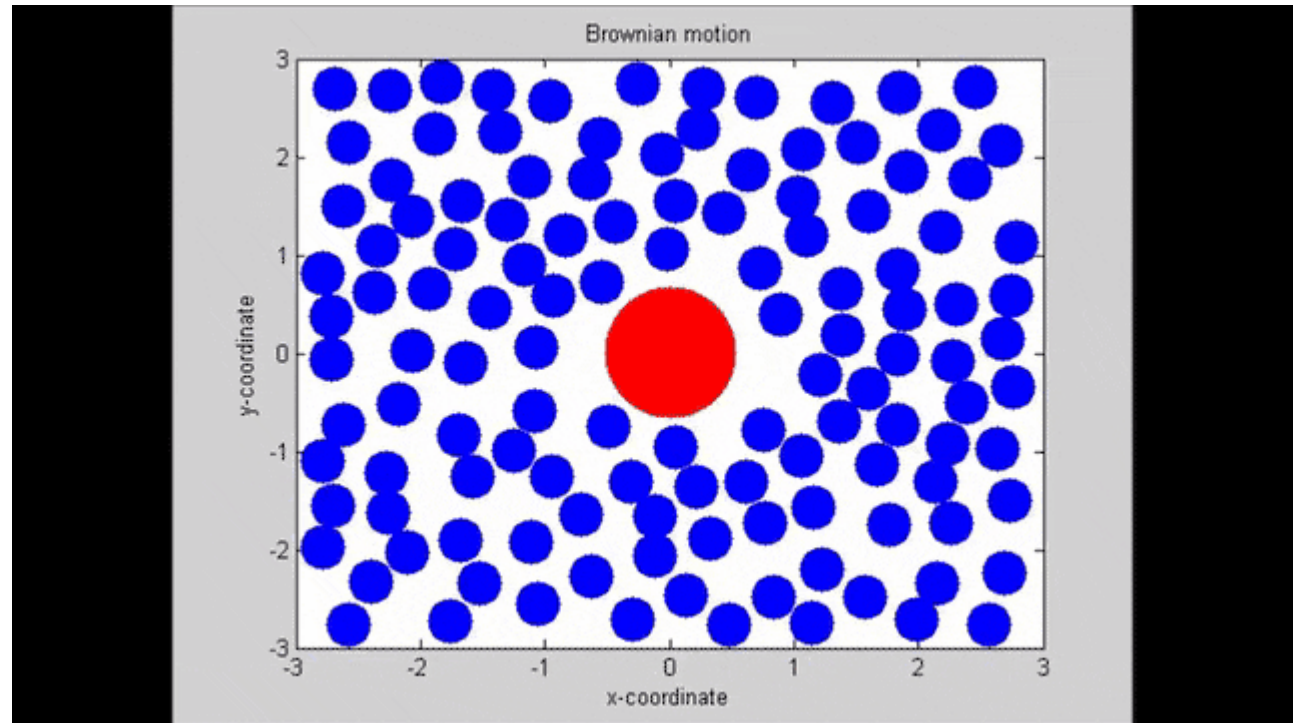
2. Temperature

- the higher the temperature, the faster is the rate of diffusion



Brownian Movement

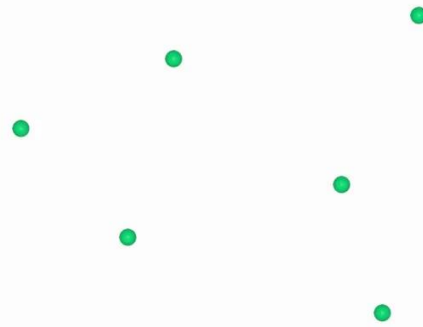
Brownian motion is the random, uncontrolled movement of particles in a fluid as they constantly collide with other particles.



Introduction – Previous Research

high concentration

low concentration



diffusion



Cold Hot

Introduction – Purpose of our research





NARROW



WIDE

Introduction - Hypothesis

Less space (Narrow) = slow rate of Diffusion

More space (Wide) = fast rate of Diffusion



Method – How to investigate.

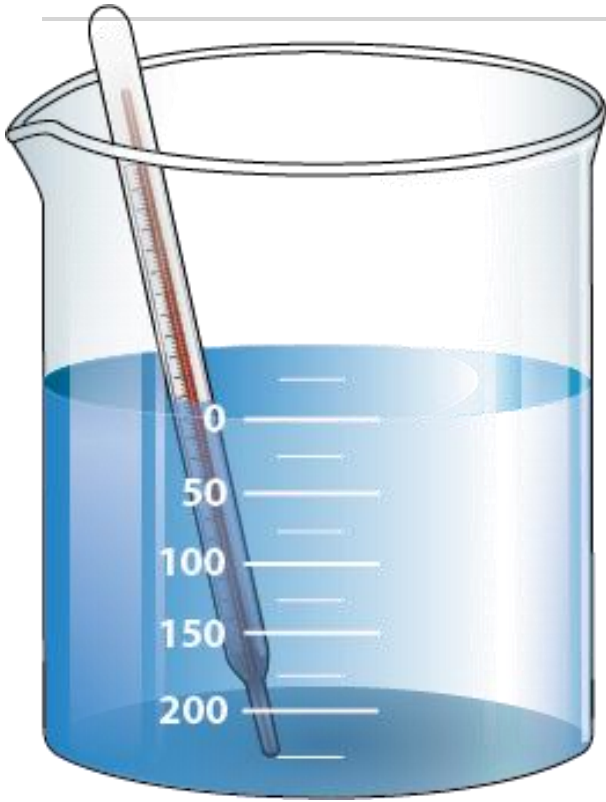


Materials

Water

Ink

Method – How to evaluate



1. Check Temperature



2. Add ink to water.



3. Record time of Diffusion.